CHEMICAL CONSTITUENTS OF GARCINIA MANGOSTANA

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Garcinia mangostana L. (Guttiferae), the Mangosteen tree, has received rather extensive investigation from various research groups. These investigations have resulted in the isolation of anthocyanin glycosides (1), a benzophenone, maclurin (2), and several xanthone derivatives (3-9), such as mangostin [1] (8). The usage of the fruit hull (pericarp) in Thai folk medicine (10) for healing skin infection and wounds and for the relief of diarrhea has led us to reexamine the plant with a view to identifying the biologically active compounds.

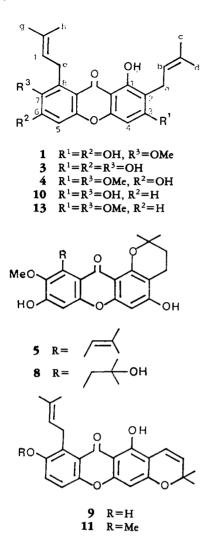
Extraction of the pericarp with C_6H_6 followed by extensive quick column chromatography of the extract has resulted in the isolation of eight xanthone derivatives. Four of the xanthones have previously been isolated: mangostin [1] (8), gartanin [2] (6), γ -mangostin [3] (7), and β -mangostin [4] (9). The other four are synthetically-known compounds but are new natural products: 1isomangostin [5] (11), 3-isomangostin [6] (11), 3-isomangostin hydrate [7] (11), and 1-isomangostin hydrate [8] (11).

Five compounds were isolated from the arils: the known compounds, mangostin [1] (8), calabaxanthone [11] (12), 2- $(\gamma, \gamma$ -dimethylallyl)-1,7-dihydroxy-3-methoxyxanthone [12] (14) and the new substances, demethylcalabaxanthone [9] and 2,8-bis- $(\gamma, \gamma$ -dimethylallyl)-1,3,7-trihydroxyxanthone [10].

Elucidation of the structures of compounds 5-12 was based largely on comparison of ¹H-nmr spectral data with those of mangostin and some of its simple derivatives. NOe enhancements

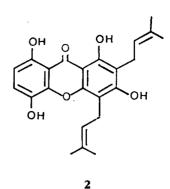
were useful for making some assignments. Use was also made of the wellknown deshielding effect of the C9 carbonyl group on protons or CH₂ groups at C8, and the fact that, generally, H4 resonates at noticeably higher field than H5. Thus, for example, in mangostin [1] the methylene group of the prenvl chain at C8 (CH₂e) resonated at δ 4.11 (cf. δ CH₂a 3.45) and showed a nOe enhancement of 3% on irradiation of the methoxyl group [other enhancements were of Hf (2,3%) and of OH (1,5%)]. With 4 irradiation of the methoxyl group δ 3.81 gave enhancements to the signals of CH_2e (2%), Hf (1.8%), and OH (1%); irradiation of the methoxyl group δ 3.90 gave strong enhancement (21%) of the signal of H4 (δ 6.35) and only weak enhancement of the signal from CH_{2a} (0.8%) (the bias in the last case was typical for the structural situation; in fact, enhancement of the neighboring CH₂ group was usually not detected at all in many compounds). In a similar fashion it was possible to assign the nmr spectra of compounds 2-13. Assignments are shown in Table 1.

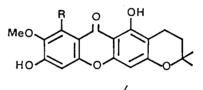
Compound 9 had the molecular formula $C_{23}H_{22}O_5$. In the ¹H-nmr spectrum H5 and H6 formed an AB pattern (δ 7.21, 7.22); H4 appeared as a fine doublet (δ 6.26) with long-range coupling (J=0.7 Hz) to Ha. The two olefinic protons in the chromene ring formed an AB quartet (δ 6.73 and 5.57), Ha and Hb, respectively, and the two methyl groups appeared together as a singlet (δ 1.47); the methylene protons in the prenyl side chain appeared as a doublet (δ 4.29), the olefinic proton as a triplet (δ 5.27), and the two methyl groups as

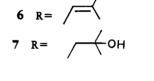


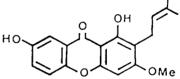
two singlets (δ 1.88 and 1.77); the C1-OH and C7-OH groups resonated at δ 13.55 and 5.37, respectively. Methylation of the compound gave calabaxanthone [11] (12), which was identified by its ¹H-nmr spectrum.

Xanthone 10 had the molecular formula $C_{23}H_{24}O_5$. The ¹H-nmr spectrum (Table 1) indicated the structure of 10 [in C₆D₆ H5, H6 resonated at δ 6.74 and 7.00, respectively, (AB quartet, J=8 Hz)]. In the ¹H-nmr spectrum $(CDCl_3-C_6D_6, 1:1)$ of the dimethyl ether derivative 13, C7-OMe (δ 3.74) gave a 12% nOe enhancement to H6 (δ 7.10, d, J=8 Hz); C3-OMe (δ 3.76) gave a 10% nOe enhancement to H4 (δ











6.33). H5 resonated at δ 7.18 (d, J=8Hz).

It is interesting that the previously reported (3) garcinones A, B, and C were not isolated in the present work. Some variability in the range of constituents must be possible.

The biological activities of these compounds will be reported elsewhere.

EXPERIMENTAL

GENERAL EXPERIMENTAL PROCEDURES.— Low and high resolution mass spectra were obtained on a AEI MS 902 instrument at 70 ev. ¹Hnmr spectra were recorded in CDCl₃ (TMS as internal reference) at 400 MHz with Bruker WM-400 instrument. NOe enhancements were measured in the difference mode by the method previ-

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TABLE

1.68 (s), 1.84 (s) 1.79 (s), 1.83 (s) 1.61(s), 1.76(s) 1.69 (s), 1.83 (s) 1.32 (s), 1.32 (s) 1.26(s), 1.26(s) 1.77 (s), 1.88 (s) 1.84 (s), 1.89 (s) 1.82 (s), 1.85 (s) 1.46 (s), 1.46 (s), 1.80 (s), 185 (s) *TMS as internal reference; CDCl₃ as solvent unless stated otherwise, s=singlet; d=doublet; t=triplet; b=broad peak; m=second order multiplet. 1.78 (s), 1.84 (s) 1.39 (s), 1.39 (s), ..37 (s), 1.37 (s), .36(s), 1.36(s), .47 (s), 1.47 (s), l. 78 (s). 1. 78 (s). 1.68 (s), 1.68 (s), l.70 (s), 1.78 (s), .64(s), 1.64(s), ..68 (s), 1.69 (s), .40 (s), 1.40 (s), 1.68 (s), 1.79 (s) c+d+g+h5.27 (bt) J=7 Hz 5.27 (bt) 5.25 (bt) . 28 (bt) =7 Hz1.79 (m) 5.30 (bt) I=7 Hz $I = 7 H_z$ $J=7 H_z$ 5.28 (bt) .34 (bt) i.28 (bt) .30 (bt) =7 Hz= 7 Hz $= 7 H_z$ (m) 67. = 7 HzI ÷ J=7 Hz*J*=7 Hz 4.16(d) 4.17 (d) =7 Hz3.36(m) 4.29 (d) $I = 7 H_z$ 3.44 (m) J=7 Hz4.27 (d) = 7 Hz4.09 (d) =7 Hz4.08 (d) 4.11(d) $= 7 H_z$ 4.11(d) =7 Hz4.19 (d) e 1.79(t) 5.57 (d) 5.23 (bt) I=7 Hz1.82 (t) $= 10 \, \text{Hz}$ $= 10 H_z$ 5.23 (bt) I=7 Hz=7 Hz1.83 (t) $I = 7 H_z$ I=7 Hz5.26(bt) $I = 7 H_z$ 5.57 (d) 5.24 (t) 1.82 (t) 5.28(t) = 7 Hz5.30(t) =7 Hz= 7 Hz=7 Hzم. 3.36(d) J=7 Hz 6.74 (dd) 5.73 (dd) 3.36(d) I=7 Hz2.64 (t) $I = 7 H_z$ =7 Hz3.35 (d) =7 Hz=7 Hz2.71(t) I = 7 Hz=7 HzJ = 10,3.45 (d) *l*= 10, 0.7 Hz = 7 Hz=7 Hz2.65 (t) 2.72(t) 0.7 Hz 3.35 (d) 3.45 (d) đ Position 9.25 (s) 3.87 (s) (OCH₃) 3.81(s) 3.81 (s) 5.37 (b) 5.45 (b) 3.87 (s) OCH₃) 2.91 (b) 3.78 (s) 3.80 (s) (OCH1) 3.85 (s) OCH₁) ("HOO 3.81(s) OCH₁) OCH₁) OCH₁) (HO) (HO) (HO) (HO) 7 =8,2 Hz 7.26 (dd) 7.21(d) $I = 8 H_z$ 7.19(s) 7.26(s) 6.35(b) 6.36(b) 7.22 (d) 7.26(s) 1.68(b) 5.33 (s) 3.31(s) 6.33 (s) 6.31(s) (HO) (HO) (HO) (HO) (HO) (HO) (HO) ы 9 7.29 (d) =8 Hz7.19(s) 7.26(s) 7.22 (d) $= 8 H_z$ 6.83 (s) 7.21(d) 7.26(s) 6.82 (s) 6.81 (s) 6.82 (s) 6.67 (s) 6.83 (s) 6.70(s) 5 \sim = 0.7 Hz= 0.7 Hz6.24 (d) 6.26(d) 6.34(s) 6.30(s) 6.41(s) 6.28 (s) 6.35(s) 6.34(s) 6.24(s) 6.34(s) 6.37 (s) 6.24 (s) 4 3.91 (s) (OCH3) 6.37(b) 3.92 (s) (OCH₁) 6.37 (s) 3.90(s) (986) 5.33 (s) 6.16(s) OCH₃) (HO)(HO) (HO) (HO) (HO) 3 (3.03 (s) 3.39(s) 13.60(s) .3.63 (s) (s) (s) (3.71(s) (3.55 (s) (3.41(s) (3.93 (s) (HO) (s) 80 (s) (HO) (HO) (HO) (HO) (HO) (HO) (HO) (HO) (HO) 1 ----Compound **12**c,d 13 10 11 ŝ ĕ

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^bIn CD₃COCD₄.

ously described (13). The ir spectra were measured on a Beckman Acculab 3 Infrared Spectrometer in KBr pellets or in Nujol mulls. The uv spectra were run on a Beckman UV/VIS Spectrometer 26. Melting points were recorded on a Buchi Tottoli melting point apparatus. Analytical tlc, preparative tlc, and quick column chromatography were performed on Si gel (Merck GF_{254}). For quick column chromatography mixtures were dissolved in the minimum amount of solvent and then adsorbed on Si gel prior to elution with solvents. Known compounds were identified by comparison of mp and spectral data with those previously reported.

PLANT MATERIAL.—Fruits of *G. mangostana* were collected in the Nakhon Si Thammarat Province, Thailand. A herbarium specimen has been deposited in the herbarium, Department of Biology, Prince of Songkla University, Thailand.

EXTRACTION OF PERICARPS OF G. MANGO-STANA.-The powdered, dry pericarp of G. mangostana (1 kg) was immersed in C₆H₆ (5 liters) for 3 days. The extraction was repeated three times, and the combined solution was concentrated to 700 ml; a yellow precipitate which formed on cooling the solution was collected (30 g). The precipitate was separated on a quick column eluting with hexane, hexane/C₆H₆, C₆H₆, CH₂Cl₂, EtOAc, Me₂CO, and then with MeOH. The eluents containing similar components were combined into seven fractions. Fractions 1,3,5 each contained one major component while fractions 2,4,6, and 7 contained mixtures of several compounds. Fraction 1 yielded gartanin [2] (1.4 g), yellow needles from hexane/C₆H₆, mp 170-172° [lit. (6) mp 167°]. Fraction 3 consisted of mangostin [1] (11.3 g), mp 180-182° [lit. (8) mp 182-183°]. Fraction 5 gave γ-mangostin [3] (1.2 g), mp 207-211° [lit. (7) mp 207°].

The mother liquors from fractions 1 and 3 were combined with fraction 2, and the material was subjected to another quick column separation. Three major fractions (tlc) were obtained: A,B, and C. Fraction B was mainly mangostin; fraction C gave β -mangostin [4] (0.03 g), mp 176-180° [lit. (9) mp 175-176°]. Fraction A yielded on fractional crystallization from C₆H₆, firstly γ mangostin (0.2 g) and then 3-isomangostin [6] (0.15 g) mp 155-160° [lit. (11) mp 154-155°]. From the mother liquors β -mangostin (0.03 g) was obtained.

ISOLATION OF 1-ISOMANGOSTIN [5], 3-ISO-MANGOSTIN HYDRATE [7], AND 1-ISO-MANGOSTIN HYDRATE [8].—Fraction 4 and 6 from the initial quick column were combined with the mother liquor of fraction 5, and the material was separated on a quick column, eluting with C_6H_6 , CH_2Cl_2 , $CH_2Cl_2/EtOAc$, then MeOH. The fractions containing similar components were combined into four fractions, D,E,F, and G. Fractions D and E contained mangostin [1] and γ -mangostin [3], respectively, as the major components. Fraction F yielded 1-isomangostin [5] (0.06 g), mp 245-249° [lit. (11) mp 241-243°]. Crystallization of Fraction G from CH₂Cl₂ gave, as a first crop, 3-isomangostin hydrate [7] (0.015 g), mp 180-182° [lit. (11) mp 182-183°] and, as a second crop, 1-isomangostin hydrate [8] (0.03 g), mp 255-257° [lit. (11) mp 261-263°].

EXTRACTION OF ARILS OF G. MANGO-STANA.-Fresh arils obtained from seeds of G. mangostana (2.7 kg) were macerated in Me₂CO. The Me₂CO extract gave on evaporation a syrup which was fractionated between a minimum amount of H₂O (150 ml) and a large quantity of EtOAc until no significant quantity of substance was obtained from the EtOAc layer. The combined EtOAc solution was evaporated to afford a syrup (6.2 g) which was separated on a quick column eluting with hexane, hexane/C₆H₆, C₆H₆, C₆H₆/CH₂Cl₂, CH₂Cl₂, then EtOAc. The fractions containing similar components were combined into seven fractions: I, J, K, L, M, N, and O. Fraction J consisted of a mixture. Each of the other fractions contained a major component. Fraction I yielded calabaxanthone [11] (0.014 g), mp 170-172° [lit. (12) mp 172°]. Fraction O was largely mangostin [1] (0.1 g), and Fraction N consisted of 2-(y, y-dimethylallyl)-1,7-dihydroxy-3-methoxyxanthone [12] (0.1 g), identical with that obtained previously (4).

ISOLATION OF DEMETHYLCALABAXANTHONE [9].—Crystallization of Fraction K from CH₂Cl₂ gave yellow needles (0.06 g) of demethylcalabaxanthone [9], mp 184-190°. Elemental analysis found: C, 72.5; H, 5.9. C₂₃H₂₂O₅ requires C, 73.0; H, 5.9%; uv λ max (EtOH) (nm, log ϵ) 225 (4.28), 235 (4.26), 286 (4.74), 386 (4.21); ir (KBr) (cm⁻¹) 3450, 1640, 1610; ms m/z (rel. int. %) M⁺⁺ 378 (15), 363 (46), 335 (13). Methylation of 9 (Me₂SO₄/NaOH, room temperature) gave calabaxanthone [11], identical with the sample previously isolated.

ISOLATION OF 2,8-BIS(γ, γ -DIMETHYLALLYL)-1,3,7-TRIHYDROXYXANTHONE **[10]**.—Fraction M was purified on ptlc, eluting with CH₂Cl₂. The substance from the major band was crystallized from C₆H₆ to give yellow prisms (0.04 g) of 2,8-bis(γ, γ -dimethylallyl)-1,3,7-trihydroxyxanthone **[10]**, mp 168-170° (Found: M^{+.} 380.1651. C₂₃H₂₄O₅ requires 380.1623); uv λ max (EtOH) (nm, log ϵ) 241(4.52), 266(4.49), 315(4.25), 379(3.72); ir (KBr) (cm⁻¹) 3390, 1635; ms m/z (rel. int. %) M^{+.} 380(100) 337(51), 324(71), 309(82). Methylation of **10** gave the dimethyl ether **13**, mp 181-183° (Found: M^{+.} 408.1970. C₂₅H₂₈O₅ requires 408.1936); ms m/z (rel. int. %) M⁺⁺ 408(100), 365(68), 352(88), 337(68), 321(19), 309(37).

ACKNOWLEDGMENTS

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